ween the two r any one of ted again by points. As the apart of the was arbitrarily ingled out. As are considered ntal points are

btained in this es reported for liquid benzene reases with the ion; at constant ne of n-heptane ropanol concenns of 1-propanol. reases with the

Properties of Aliphatic and Aromatic Aldehydes under High Pressure

Compressibility and Viscosity Determination

P. MITRA CHAUDHURI1, R. A. STAGER, and G. P. MATHUR Department of Chemical Engineering, University of Windsor, Windsor, Ontario, Canada

> Compressibility and viscosity measurements have been made on seven aliphatic and three aromatic aldehydes at pressures up to 20,000 p.s.i.g. The viscosity data have been correlated by means of an empirical equation. Experimental data are presented in graphical form for the pressure range investigated.

DATA on the viscosity and compressibility of liquids are generally scarce. Previous investigations have concentrated mainly on organic compounds. The work on high pressure physical properties was pioneered by Bridgman

Almost no work has so far been reported on aldehydes, possibly because they are difficult to work with. The aliphatic aldehydes are low boiling liquids with a disagreeable smell and are harmful to the human system. The aromatic aldehydes have a tendency to be oxidized to their corrent in the vapor phas sponding acids.

and, Mich., 1953.

Chem. Eng. Sci. 20,

terdam, 1950.

EXPERIMENTAL

Concurrent measurements of the physical properties were conducted in a compact apparatus (3). Compressibility was measured by a piston displacement device (2) in conjunction with viscosity measurement, which was performed by a falling cylinder method (4, 5).

Pressure was generated in the equipment by means of arnegie Inst. Technol manually operated hydraulic pump. The pressure was neasured by a Bourdon gage. The precision of pressure J. Sci. Res. Ser. A neasurement was within $\pm 0.2\%$. The temperature was J. Chem. Eng. Days F.

The compressibility meter consisted of a 16-inch long, Thermodynamics," 4-inch I.D. stainless steel tube with cap-collar-gland asoperties of Chemicembly at both ends. Inside the tube there was a pistonhagnet assembly soldered together. Outside the comhem. (London) 16, 3 ressibility meter was a magnet-pointer assembly. The ointer indicated the position of the piston to one tenth of H. T., Proc. Am. Accmillimeter.

The viscometer consisted of a fall tube, 152.5294 cm. J. Chem. Eng. 37, Ing with 0.4710 cm. inside diameter. The tube was of ainless steel. The plummet used was a thin hollow cylinder f a uniform outside diameter of 0.43688 cm. with hemical Constants of pherical ends. Bridgman had used plummets with three rojecting lugs to keep the plummet coaxial during its nde, A. K., Lu, B. C. III. The lugs introduced some error in the measurements ad were omitted in the present setup. Modified electrical intacts, with specially designed plugs, made the use of gs unnecessary. The coaxial fall of the plummet through ccepted October 24, 19e viscometer tube was confirmed by means of a stethoesearch Council of Canope.

ETHOD OF OPERATION

Before starting experiments with the aldehydes, several st runs were made with various liquids of known viscosity examine the accuracy of the apparatus. Plummets of fferent dimensions were tried to find the optimum plumt size. The authors discovered that plummets with esent address: General Electric Co., Guelph, Ontario, Canada

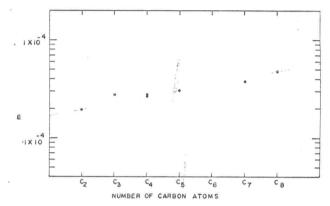
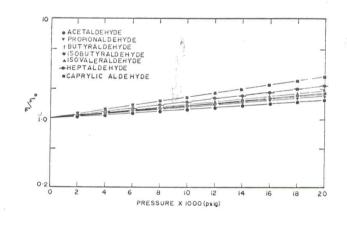


Figure 1. m against number of carbon atoms



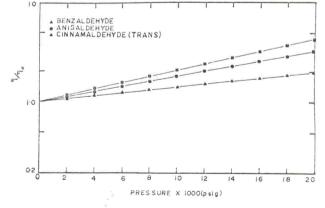


Figure 2. Viscosity-pressure diagrams